

Natural units

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- Normal matter

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- Atomic units

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- Atomic units
- SI to AU conversion

- normal matter is governed by electrostatic forces and non-relativistic quantum mechanics

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- typical values for normal matter are reasonable numbers
- times products of powers of these constants

Constants of nature

Natural units

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examples

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- the powers of SI units

	m	kg	s	A
\hbar	2	1	-1	0
e	0	0	1	1
m	0	1	0	0
$4\pi\epsilon_0$	-3	-1	4	2

Powers of natural constants

Natural units

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- read: s = number multiplied by $\hbar^3 e^{-4} m^{-1} (4\pi\epsilon_0)^2$ etc.
- find out number and exponents for arbitrary SI unit

```
function [value,power]=atomic_unit(si)
val=[1.05457e-34,1.60218e-19,9.10938e-31,4*pi*8.85419
dim=[2 1 -1 0; 0 0 1 1; 0 1 0 0; -3 -1 4 2];
mid=round(inv(dim));
power=si*mid;
value=prod(val.^power);
```

```
>> length = [0 1 0 0];
>> length = [1 0 0 0];
>> [astar,apow]=atomic_unit(length);
>> astar
    5.2917e-11
>> apow
     2     -2     -1     1
```

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e. g. atomic length unit

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- i. e. Bohr's radius

$$a_* = \frac{4\pi\epsilon_0\hbar^2}{me^2} = 5.2917 \times 10^{-11} \text{ m}$$

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$$-\frac{\hbar^2}{2m}\Delta\phi + \frac{1}{4\pi\epsilon_0}\frac{-e^2}{r}\phi = E\phi$$

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Electric field strength

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Magnetic induction

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- mass is mass of nucleons, therefore must be multiplied by approximately 4000

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- relativistic corrections

Pockels coefficients

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- lithium niobate is rather resilient to electrical fields

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- steel : $E = 200 \text{ GPa}$

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- $N \approx 1$ and $\Omega \approx 1$ in natural units!